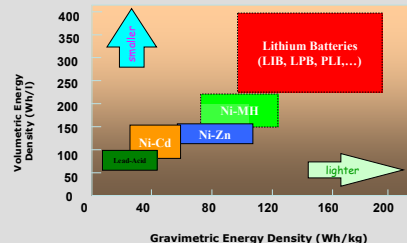


Lithium Batteries Offer Best Performance



Development of Advanced Materials

Anode Materials

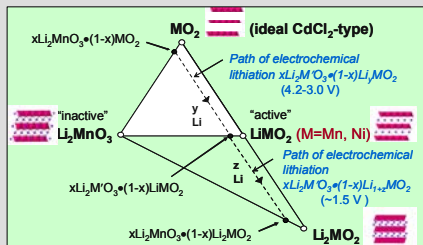
- Lithiated intermetallic anodes
- Carbon-coated nanophase lithium titanate anodes

Cathode Materials

- Lithium-rich composite cathodes (w/high Mn content)
- Surface-treated lithium metal oxide cathodes
- Surface-coated lithium metal oxide cathodes
- Carbon-coated lithium iron phosphate

Electrolyte Components & Systems

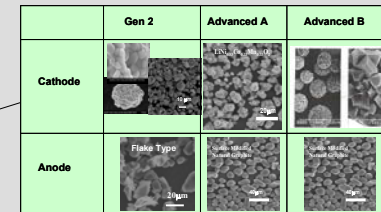
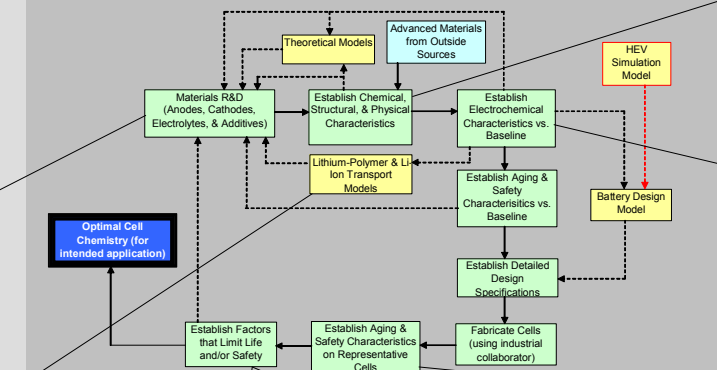
- Additives that form more stable SEI layers
- Additives as flame retardants for organic carbonates
- Additives that establish redox shuttle & prevent cell overcharge
- More stable & non-flammable solvent systems



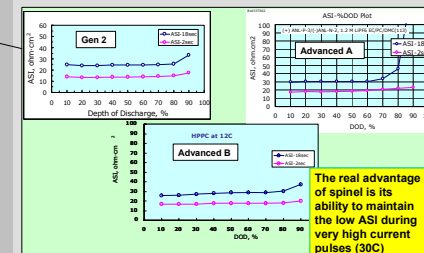
Argonne has a large portfolio of intellectual property covering its advanced materials, including patents on a family of composite cathode materials.

Argonne's Battery R&D Process

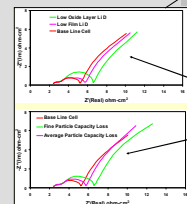
Advanced materials & optimal cell chemistries are developed in the context of understanding the fundamental limitations of existing materials. This flow diagram illustrates the process used to develop more optimal materials and cell chemistries for hybrid electric vehicle batteries.



SEM images of alternative anode and cathode materials for use in high-power Li-Ion batteries for hybrid electric vehicle applications. The physical and chemical properties of these materials control their performance in Li-Ion cells. Spherical round edge particles of the proper size are desired for optimal performance, life, and safety.



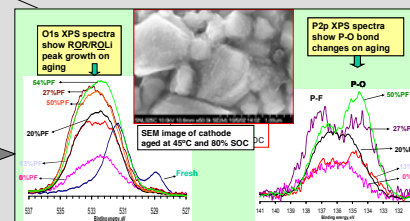
Area specific impedance values measured, using FreedomCAR test protocols, on cells containing alternative cathode materials show their ability to meet performance requirements.



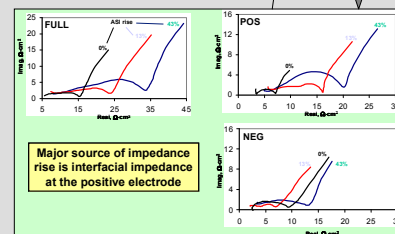
Electrochemical model indicates that 2 most probable mechanisms for explaining positive electrode EIS data are:

1. Reduced interfacial diffusion due to changes in surface films or in oxide surface layer
2. Preferential isolation of small active material particles

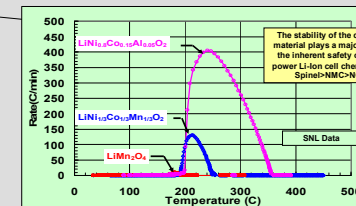
Electrochemical model is used to validate the significance of observed physical & chemical phenomena.



XPS and SEM data on harvested cathodes show a film on the surface of the cathodes & its chemical composition changes as the cells age.



Electrochemical impedance data on electrodes harvested from aged cells show that the main source of impedance rise is interfacial impedance at the cathodes.



Comparative accelerated rate calorimetry (ARC) tests show the reduced thermal reactivity of cells that employ alternative cathodes to our Gen 2 cathode material.

